# Synthesis and pharmacological evaluation of derivatives structurally related to nimesulide

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Summary— The present work reports the synthesis of a series of compounds structurally related to the antiinflammatory and antihistaminic agent nimesulide (I), in which the *p*-nitrophenyl moiety has been replaced by pyridine (1a-c) and pyridine *N*-oxide (2a-c). In addition, two compounds (3a, 4a) have been synthesized in which the *p*-nitro group of I was substituted by a cyano and a 1*H*-tetrazol-5-yl group, respectively. Representative 1a and 2a were also modified by replacing the methanesulfonamido group with an acetamido group (5a, 6a). The pharmacological evaluation of compounds 1-6 in comparison to I, indicates that such modifications are detrimental to the activity. Moreover 3a and 4a caused bronchoconstriction and hypotension, thus behaving as histaminic-like rather then antihistaminic agents.

nimesulide / antiinflammatory-antihistaminic activity / N-pyridinyl analogue

#### Introduction

I

The non-steroidal antiinflammatory analgesic agent nimesulide I is currently employed in the treatment of inflammatory conditions, pain and fever states. This compound also displays both antianaphylactic and antihistaminic activity; in comparison, indometacin considerably potentiates histamine release, in spite of its inhibitory potency with regard to tromboxane  $A_2$  formation [1].

Recently, compound CGP-28237 II, structurally related to I, was reported to display antiinflammatory properties with excellent gastrointestinal tolerability [2].

The positive results of the bioisosteric approach which led to **II**, prompted us to synthesize analogues

П

of nimesulide in which the p-nitrophenyl moiety was replaced by pyridine (1a–c) and pyridine N-oxide (2a–c), as well as by 4-cyano (3a) and 4-(1H-tetrazol-5-yl)phenyl groups (4a). In addition compounds 5a and 6a have been prepared in which the methanesulfonamido group of 1a and respectively, 2a is replaced by an acetamido group (see fig 1).

# Chemistry

The synthesis of 1 and 2 (scheme 1) was performed through the common intermediate 3-bromo-4-nitropyridine-N-oxide 8 in turn obtained by conversion of 3-bromopyridine with peracetic acid into its N-oxide 7, followed by nitration with HNO<sub>3</sub>/H<sub>2</sub>SO<sub>4</sub> at 90 °C.

Heating of 8 with phenol or *p*-substituted phenols in *tert*-butanol, in the presence of potassium *tert*-butoxide, gave 60–70% of the aryloxy derivatives **9a**-**c** besides minor amounts (30%) of 4-aryloxy-3-bromopyridine-*N*-oxides **11a**-**c**, easily separated by silica gel column chromatography. Reduction of **9a**-**c** with powdered iron in acetic acid to 4-aminopyridine derivatives **10a**-**c**, and subsequent condensation with mesyl chloride in 10% sodium hydroxide, afforded **1a**-**c** in 55–60% yield; these were then converted by peracetic acid to the *N*-oxides **2a**-**c** in 40–67% yields.

To obtain 4-cyano- and 4-(1-H-tetrazol-5-yl)-2-phenoxymethanesulfonanilide **3a**, **4a**, nimesulide **I** was reduced by catalytic hydrogenation, yielding 97% of **12a**, which was diazotized and added to a solution of CuCN and NaCN to give 72% of **3a**. The reaction of **3a** with trimethyltin azide in refluxing toluene, followed by destannylation with silica gel, afforded **4a** in 33% yield (see scheme 2). Finally the 4-acetamido derivatives **5a** and **6a** (scheme 3) were prepared from **10a** and its *N*-oxide **13a**, in turn obtained by catalytic hydrogenation of **9a**, by refluxing with acetic anhydride in acetonitrile. It is of note that **13a** failed to be converted into **2a** with mesyl chloride in 10% sodium hydroxide.

#### Results and discussion

The antihistaminic activity of compounds 1–6 and the antiinflammatory activity of compounds 1a–4a were evaluated in vivo, as described in the *Experimental protocols*.

# Antihistaminic activity

None of the compounds (1a-c, 2a-c, 3a-6a) displayed significant protection against histamine-induced bronchoconstriction. In comparison nimesulide reduced by 89% the effect of this autacoid on the airways, whereas indomethacin was devoid of this activity.

In contrast, when compounds 3a and 4a were injected intravenously at a dose of 10 mg/kg in anaesthetized guinea pig, a prompt increase in intratracheal pressure (ITP) and a fall in blood systemic pressure were recorded (fig 2). Both bronchoconstriction and hypotension, caused by two of the compounds under examination, 3a and 4a, were in the potency range of that of 5–10 µg/kg of histamine. The bronchoconstrictive activity of 3a and 4a was only partially reversible, since 5 min after their administration the ITP values were still twice the basal values. Moreover the increase in bronchial resistance caused by compounds 3a and 4a is abolished by pre-treatment of the animals with pyrilamine (PYR) at a dose of 2 mg/kg iv (see fig 2). This finding suggests that the action of the above compounds involves stimulation of H<sub>1</sub>-histamine receptors in the airways.

# Antiinflammatory activity

The aim of these experiments was to evaluate the antiinflammatory activity of representative compounds **1a–4a** in comparison to nimesulide. In the test used (carrageenin edema of the rat) nimesulide has been reported to have an ED<sub>50</sub> of 1.25 mg/kg os [3].

Fig 1. Compounds 1a-c, 2a-c, 3-6.

Therefore in order to evidence possible antiinflammatory effects in the new compounds, even if they are lower than those of the reference drug, a dose of 10 mg/kg per os of 1a—4a was administered to the test animal. All the tested compounds were found to be devoid of antiinflammatory activity. We note that a US patent [4] appeared while this work was in progress which claimed, for a class of substitutued phenoxypyridines including 5a, 9a and 10a, a topical antiinflammatory property in mice against ear edema induced by TPA (phorbol 12-myristate acetate), at a dose of  $10 \,\mu\text{g/kg}$ . Moreover compound 3a was described in a Japanese Patent as an antiinflammatory agent in carrageenin edema of the rat at a dose of  $50 \, \text{mg/kg}$  [5].

# **Experimental protocols**

#### Chemical synthesis

Melting points were determined on a Büchi 510 capillary apparatus and are uncorrected. Analyses indicated by the chemical symbols were within ±0.4% of the theoretical values. <sup>1</sup>H-NMR spectra were recorded on a Bruker AC 200 spectro-

a) H<sub>2</sub>O<sub>2</sub>, AcOH, 70-80°C, 7h b) HNO<sub>3</sub>, H<sub>2</sub>SO<sub>4</sub>, 90°C, 1.5 h

c) R—OH t-BuOK, t-BuOH, 12h d) Fe, AcOH, 60-70°C, 1h e) CH3SO<sub>2</sub>Cl, H<sub>2</sub>O, 2,5 M NaOH, 0-3°C. f) H<sub>2</sub>O<sub>2</sub>, AcOH, 70-80°C, 7h

Scheme 1. Synthesis of compounds 1a-c, 2a-c.

2a-c

meter; chemical shifts are reported as ppm, relative to tetramethylsilane as an internal standard. TLC on silica gel plates (Merck) was used to check product purity. Silica gel 60 (Merck; 70-230 mesh) was used for column chromatography. The structures of all compounds were consistent with their analytical and spectroscopic data.

# 3-Bromo-4-nitropyridine-N-oxide 8

The method described by Ochiai [6] was modified as follows. To a solution of 32 g (0.202 mol) of 3-bromopyridine in 120 mL CH<sub>3</sub>COOH, 41 mL 30% H<sub>2</sub>O<sub>2</sub> (0.404 mol) was added, and the solution was heated for 9 h at 70–80 °C, then kept overnight at room temperature. After concentration under reduced pressure, the residue (about 40 mL) was made alkaline with excess solid Na<sub>2</sub>CO<sub>3</sub> and diluted with 100 mL CHCl<sub>3</sub> under stirring. The inorganic salts were filtered off, the organic layer was dried (Na<sub>2</sub>SO<sub>4</sub>) and the solvent evaporated in vacuo. The residue was distilled, collecting 31.9 g (90%) of 3-bromopyridine-N-oxide 7 (bp: 100 °C/0.5 mmHg). This compound was dissolved in 77 mL H<sub>2</sub>SO<sub>4</sub> on cooling, and to the resulting solution,

kept at 0–5 °C, a mixture of 128 mL conc HNO<sub>3</sub> and 77 mL conc  $\rm H_2SO_4$  was cautiously added. After heating at 90 °C for 1.5 h, the reaction mixture was poured onto ice and the pH adjusted to 8 with 50% NaOH. The yellow solid which separated was filtered and triturated with CHCl<sub>3</sub>. Filtration and evaporation of the solvent gave 25.3 g of 8, mp: 152 °C (lit [6] mp: 159 °C). An additional 3 g of 8 were recovered from the alkaline filtrate by extraction with CHCl<sub>3</sub>, drying of the organic layer, evaporation of the solvent and crystallization of the crude residue (4 g) from benzene. Yield 72%. <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 7.9–8.4 (m, 2H), 8.6 (s, 1H, H-2).

4-Nitro-3-phenoxypyridine-N-oxide **9a** and 3-bromo-4-phenoxypyridine-N-oxide **11a** 

To a solution of phenol (0.6 g, 6.38 mmol), t-BuOK (0.51 g, 4.57 mmol) and 15 mL tert-butanol (1 g, 4.57 mmol) of 8 was added portionwise. After refluxing for 16 h, the reaction mixture was cooled and poured into ice under stirring. The solution was made alkaline with 10% NaOH and extracted with 3 × 15 mL CHCl<sub>3</sub>. The organic layers were collected and dried (Na<sub>2</sub>SO<sub>4</sub>), and the solvent was evaporated to give a yellow solid which was fractionated by column chromatography (silica gel, CH<sub>2</sub>Cl<sub>2</sub>/MeOH 95:5) collecting respectively 0.65 g (61%) of 9a and 0.35 g (29%) of 11a.

9a: mp: 110 °C (cyclohexane/benzene, 1:1) (lit [4] mp:

**9a**: mp:  $110^{\circ}$ C (cyclohexane/benzene, 1:1) (lit [4] mp: 110-111, EtOH),  ${}^{1}$ H-NMR (CDCl<sub>3</sub>)  $\delta$ : 7.1-7.8 (m,  $5H_{arom}$ ), 7.8-8.2 (m, 3H).

**11a**: mp: 154–155 °C (cyclohexane/benzene, 1:1), <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 6.5–7.0 (d, 1H, J = 8 Hz, H-5), 7.0–7.8 (m, 5H<sub>arom</sub>), 7.9–8.3 (d, 1H, J = 8 Hz, H-6), 8.6 (s, 1H, H-2).

#### Similarly prepared were:

OCH3 : b

CI: c

3-(4-Methoxyphenoxy)-4-nitropyridine-N-oxide **9b**. Yield 68%; mp: 111-112 °C,  $^{1}$ H-NMR (CDCl<sub>3</sub>)  $\delta$ : 3.8 (s, 3H, OCH<sub>3</sub>), 6.8–7.4 (m, 4H<sub>aron</sub>), 7.6–8.3 (m, 3H).

3-Bromo-4-(4-methoxyphenoxy)pyridine-N-oxide 11b. Yield 30%; mp: 178 °C,  $^{1}$ H-NMR (CDCl<sub>3</sub>)  $\delta$ : 3.8 (s, 3H, OCH<sub>3</sub>), 6.5–6.6 (d, 1H, J = 8 Hz, H-5), 6.8–7.1 (m, 4H), 7.8–8.1 (d, 1H, J = 8 Hz, H-6), 8.4 (s, 1H, H-2).

3-(4-Chlorophenoxy)-4-nitropyridine-N-oxide 9c. Yield 64%; mp: 108 °C, ¹H-NMR (CDCl<sub>3</sub>)  $\delta$ : 6.8–7.6 (m, 4H<sub>arom</sub>), 7.6–8.1 (m, 3H).

3-Bromo-4-(4-chlorophenoxy)pyridine-N-oxide 11c. Yield 35%; mp: 144–145 °C, ¹H-NMR (CDCl<sub>3</sub>)  $\delta$ : 6.65 (d, 1H, J = 8 Hz, H-5), 7.0–7.1 (m, 4H), 8.0–8.1 (m, 1H, H-6), 8.4 (s, 1H, H-2).

# 4-Amino-3-phenoxypyridine 10a

To a stirred solution of 2.7 g (11.6 mmol) of  $\bf 9a$  in 54 mL CH<sub>3</sub>COOH, 4.2 g (75.2 mmol) powdered Fe was added in small portions, keeping the temperature at 60 °C. At the end of the addition the mixture was stirred for an additional 30 min, then the insoluble material was filtered and washed with acetic acid. The filtrate was evaporated in vacuo to give a brown oil which was treated with 10 mL H<sub>2</sub>O, the pH adjusted to 10 with aq NaOH, and extracted with 3 × 30 mL CHCl<sub>3</sub>. The organic layers were collected, dried over Na<sub>2</sub>SO<sub>4</sub> and the solvent was evaporated to give 1.97 g (91%) of  $\bf 10a$  as a yellow solid, mp 60–62 °C. The analytical sample was purified by column chromatography (silica gel, CH<sub>2</sub>Cl<sub>2</sub>/MeOH, 95:5), mp: 68–70 °C. ¹H-NMR (CDCl<sub>3</sub>)  $\delta$ : 4.3 (D<sub>2</sub>O exchangeable s, 2H, NH<sub>2</sub>), 6.7 (d, 1H, J = 6 Hz, H-5), 6.9–7.5 (m, 5H<sub>arom</sub>), 8.0–8.2 (m, 2H).

- a) H2/10% Pd/C, EtOH
- b) 1) NaNO2/HCI, 5°C
- 2) CuCN, NaCN, H2O, benzene, 40°C, 1h
- c) 1) Me<sub>3</sub>SnN<sub>3</sub>, toluene, 24h
- 2) silica gel, MeOH, 4h

Scheme 2. Synthesis of compounds 3a and 4a.

# Similarly prepared were:

4-Amino-3-(4-methoxyphenoxy)pyridine **10b**. Yield 60%; mp: 101-102 °C.  $^1$ H-NMR (CDCl<sub>3</sub>)  $\delta$ : 3.8 (s, 3H, OCH<sub>3</sub>), 4.4 (D<sub>2</sub>O exchangeable s, 2H, NH<sub>2</sub>), 6.4-7.4 (m, 5H), 7.7-8.2 (m, 2H).

4-Amino-3-(4-chlorophenoxy)pyridine 10c. Yield 62%; yellow oil. The hydrochloride, mp: 245–250 °C.  $^{1}$ H-NMR ( $d_6$ -DMSO)  $\delta$ : 7.0–7.8 (m, 5H), 7.8–8.8 (m, 4H), 13.9 ( $D_2$ O exchangeable s, 1H, hydrochloride)).

4-Methanesulfonamido-3-phenoxypyridine 1a

To a vigorously stirred mixture at 0–3 °C of 1.19 g (6.39 mmol) of **10a** in 10 mL Et<sub>2</sub>O, and H<sub>2</sub>O basified with a few drops of 1 M NaOH, 2.97 mL (38.3 mmol) of CH<sub>3</sub>SO<sub>2</sub>Cl were added dropwise, under cooling, keeping the mixture alkaline by simultaneous addition of 10% NaOH. After additional stirring at room temperature, the organic layer containing unreacted **10a** was separated and the aqueous layer was adjusted to pH 7–8 with aq HCl to separate 1 g (59%) of **1a** as a white solid, mp: 220–221 °C (CH<sub>3</sub>COOH/H<sub>2</sub>O, 8:2).

<sup>1</sup>H-NMR ( $d_6$ -DMSO) δ: 2.8 (s, 3H, CH<sub>3</sub>), 7.0 (d, 2H, J = 8 Hz, H-2', H-6'), 7.1 (t, 1H, J = 8 Hz, H-4'), 7.3–7.6 (m, 3H, H-5, H-3', H-5'), 8.0–8.3 (m, 2H, H-2, H-6). Anal C<sub>12</sub>H<sub>12</sub>N<sub>2</sub>O<sub>3</sub>S (C, H, N, S).

#### Similarly prepared were:

4-Methanesulfonamido-3-(4-methoxyphenoxy)pyridine **1b**. Yield 54%, mp: 187 °C (CH<sub>3</sub>COOH/H<sub>2</sub>O, 1:3), <sup>1</sup>H-NMR ( $d_6$ -DMSO) δ: 2.9 (s, 3H, CH<sub>3</sub>), 3.8 (s, 3H, OCH<sub>3</sub>), 6.9–7.1 (m, 4H<sub>arom</sub>), 7.4 (d, 1H, J = 8 Hz, H-5), 7.9 (s, 1H, H-2), 8.0 (d, 1H, J = 8 Hz, H-6). Anal C<sub>13</sub>H<sub>14</sub>N<sub>2</sub>O<sub>4</sub>S (C, H, N, S).

Scheme 3. Synthesis of compounds 5a and 6a. a) (CH<sub>3</sub>CO)<sub>2</sub>O/CH<sub>3</sub>CN; b) H<sub>2</sub>/10% Pd-C/EtOH.

b) H<sub>2</sub>/10% Pd-C/EtOH

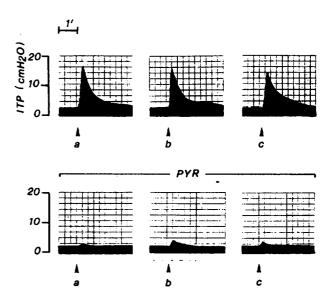


Fig 4. Original tracing showing the bronchoconstrictive effects of histamine (a;  $10 \mu g/kg$ ) iv), compound 3a (b;  $10 \mu g/kg$  iv) and compound 4a (c;  $10 \mu g/kg$  iv), before and after treatment with pyrilamine (PYR;  $2 \mu g/kg$  iv) in anaesthesized guinea pigs. Pyrilamine was administered 5 min before the agonists.

4-Methanesulfonamido-3-(4-chlorophenoxy)pyridine 1c. Yield 57%, mp: 217 °C (CH<sub>3</sub>COOH/H<sub>2</sub>O, 1:1).  $^{1}$ H-NMR ( $d_6$ -DMSO) 8: 2.8 (s, 3H, OCH<sub>3</sub>), 6.9–7.1 (m, 2H), 7.3–7.6 (m, 3H), 8.0–8.1 (d, 1H, J = 9 Hz, H-6), 8.2 (s, 1H, H-2). Anal  $C_{12}H_{11}CIN_2O_3S$  (C, H, Cl, N, S).

4-Methanesulfonamido-3-phenoxypyridine-N-oxide 2a A solution of 0.5 g (1.89 mmol) of 1a in 7.5 mL CH<sub>3</sub>COOH and 0.38 mL (3.74 mmol) of 30% H<sub>2</sub>O<sub>2</sub> was heated at 70–80 °C for 8 h. After concentration in vacuo, the oily residue was diluted with EtOH and unreacted 1a was removed by filtration. The filtrate was evaporated and the residue was treated with H<sub>2</sub>O to separate 0.25 g (47%) of 2a as a pink solid, mp: 150–151 °C. The hydrochloride melted at 145 °C (EtOH). <sup>1</sup>H-NMR (hydrochloride) ( $d_6$ -DMSO)  $\delta$ : 3.2 (s, 3H, CH<sub>3</sub>), 4.4 (D<sub>2</sub>O exchangeable broad s, 1H, NH), 7.1–7.4 (m, 3H), 7.4–7.6 (m, 2H), 7.7 (d, 1H, J = 7 Hz, H-5), 8.3 (s, 1H, H-2), 8.5 (d, 1H, J = 7 Hz, H-6). Anal (hydrochloride)  $C_{12}H_{13}CIN_2O_4S$  (C, H, Cl, N, S).

4-Methanesulfonamido-3-(4-methoxyphenoxy)pyridine-N-oxide **2b** 

This was prepared as reported for **2a**. The oily reaction product was treated with EtOH (1:4), and the solid which slowly separated was purified by column chromatography (silica gel, CH<sub>2</sub>Cl<sub>2</sub>/MeOH, 95:5) to give 0.55 g (40%) of **2b**, mp: 104 °C (EtOH). <sup>1</sup>H-NMR ( $d_6$ -DMSO)  $\delta$ : 3.2 (s, 3H, CH<sub>3</sub>), 3.8 (s, 3H, OCH<sub>3</sub>), 7.0 (d, 2H, J = 8 Hz), 7.1 (d, 2H, J = 8 Hz), 7.4 (d, 1H, J = 7 Hz, H-5), 7.6 (s, 1H, H-2), 8.0 (d, 1H, J = 7 Hz, H-6), 10.4 (D<sub>2</sub>O exchangeable broad s, 1H, NH). Anal C<sub>13</sub>H<sub>14</sub>N<sub>2</sub>O<sub>5</sub>S (C, H, N, S).

4-Methanesulfonamido-3-(4-chlorophenoxy)pyridine-N-oxide 2c

A solution of 1.85 g (6.19 mmol) of 1c in 28 mL CH<sub>3</sub>COOH and 1.26 mL (12.38 mmol) 30%  $\rm H_2O_2$  was heated at 80 °C for 7 h, then kept overnight at room temperature. The precipitate was filtered and washed with Et<sub>2</sub>O to give 0.6 g of 2c. The filtrate was evaporated and the residue, treated with EtOH, separated a solid which after purification by column chromatography (silica gel, CH<sub>2</sub>Cl<sub>2</sub>/MeOH, 95:5), gave an additional 0.7 g of 2c (yield 67%). The analytical sample melted at 186–187 °C (EtOH).  $^{1}$ H-NMR ( $d_6$ -DMSO)  $\delta$ : 3.05 (s, 3H, CH<sub>3</sub>), 7.1 (d, 2H, J = 8 Hz), 7.3–7.7 (m, 3H), 7.8–8.2 (m, 2H), 10.8 (D<sub>2</sub>O exchangeable broad s, 1H, NH). Anal  $C_{12}H_{11}N_2O_4S$  (C, H, N, S).

4-Methanesulfonamido-3-phenoxyaniline 12a

A solution of 4 g (0.013 mol) of nimesulide (I) in 40 mL EtOH was hydrogenated at room temperature in the presence of 0.4 g 10% Pd/C. When hydrogen absorption ceased, the catalyst was filtered off and the filtrate was evaporated in vacuo to give 3.53 g of 12a. Yield 97%; mp: 164-165 °C. <sup>1</sup>H-NMR ( $d_6$ -DMSO)  $\delta$ : 2.9 (s, 3H, CH<sub>3</sub>), 5.3 (D<sub>2</sub>O exchangeable s, 2H, NH<sub>2</sub>), 6.1 (d, 1H, J = 3 Hz), 6.2–6.4 (dd, 1H, J = 3 Hz, J = 6 Hz), 6.9–7.3 (m, 4H), 7.3–7.6 (m, 2H), 8.8 (D<sub>2</sub>O exchangeable s, 1H, NH).

4-Methanesulfonamido-3-phenoxybenzonitrile 3a

To a cooled solution of 12a (2.0 g, 7.2 mmol) in 1.8 mL conc HCl and 10 mL  $_2$ O, 0.5 g NaNO $_2$  (0.5 g, 7.2 mmol) in 1.5 mL  $_2$ O were added dropwise, keeping the temperature under 5 °C. The resulting dense suspension was adjusted to pH 5 with Na $_2$ CO $_3$  and the mixture was added portionwise to a solution of CuCN (0.8 g, 8.9 mmol) and NaCN (0.7 g, 14.2 mmol) in 8 mL  $_2$ O and 15 mL benzene kept under vigorous stirring at 0–5 °C. After 30 min the mixture was heated at 40 °C for an

additional 1 h, then cooled and extracted with  $3 \times 15$  mL benzene. The organic layers were collected and dried (Na<sub>2</sub>SO<sub>4</sub>), and the solvent was evaporated to give a brown oil, which was purified by column chromatography (silica gel, cyclohexane) ethyl acetate, 7:3) yielding 1.5 g (72%) of 3a, mp: 135–136 °C (*i*-PrOH). <sup>1</sup>H-NMR (CDCl<sub>3</sub>) &: 1.6 (D<sub>2</sub>O exchangeable s, 1H, NH), 3.2 (s, 3H, CH<sub>3</sub>), 6.9–7.2 (m, 3H), 7.2–7.6 (m, 4H), 7.7 (d, 1H, J = 7 Hz). Anal  $C_{14}H_{12}N_2O_3S$  (C, H, N, S).

5-(4-Methanesulfonamido-3-phenoxyphenyl)-1H-tetrazole 4a Following a reported procedure [7] a mixture of 3a (0.4 g, 1.4 mmol) and trimethyltinazide [8] (1.0 g, 5.0 mmol) in 4 mL toluene, was refluxed for 24 h. After evaporation of the solvent the oily residue was dissolved in 12 mL MeOH, and treated with 4 g silica gel. After stirring at room temperature for 4h the solvent was evaporated and the residue was purified by column chromatography (silica gel, CH<sub>2</sub>Cl<sub>2</sub>/MeOH, 95:5), yielding 0.15 g (33%) of 4a as white solid, mp: 176–177 °C. ¹H-NMR (d<sub>6</sub>-DMSO) δ: 3.1 (s, 3H), 7.0–7.3 (m, 3H), 7.3–7.7 (m, 3H), 7.7–7.8 (d, 1H, 8 Hz), 7.8–8.0 (d, 1H, 8 Hz), 9.8 (D<sub>2</sub>O exchangeable s, 1H), 16.6–17.0 (D<sub>2</sub>O exchangeable broad s, 1H). Anal  $C_{14}H_{13}N_5O_3S$  (C, H, N, S).

4-Acetamido-3-phenoxypyridine 5a

A solution of 1.34 g (7.3 mmol) of **10a**, 0.74 mL (7.3 mmol) of (CH<sub>3</sub>CO)<sub>2</sub>O and 15 mL CH<sub>3</sub>CN was stirred at room temperature for 15 h. After evaporation of the solvent, the residue was dissolved in CH<sub>2</sub>Cl<sub>2</sub> and washed with 10% Na<sub>2</sub>CO<sub>3</sub>. The organic layer, dried (Na<sub>2</sub>SO<sub>4</sub>) and evaporated, gave an oily residue which was purified by column chromatography (silica gel, CH<sub>2</sub>Cl<sub>2</sub>/MeOH, 95:5) yielding 1.25 g (75%) of **5a**, mp: 132 °C (*i*-PrOH) (lit [4], mp: 128–130 °C). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 2.2 (s, 3H, CH<sub>3</sub>), 7.0–7.5 (m, 5H<sub>arom</sub>), 7.9 (D<sub>2</sub>O exchangeable s, 1H, NH), 8.1 (s, 1H, H-2), 8.25 (d, 1H, J = 6 Hz), 8.35 (d, 1H, J = 6 Hz). Anal C<sub>13</sub>H<sub>12</sub>N<sub>2</sub>O<sub>2</sub> (C, H, N).

4-Acetamido-3-phenoxypyridine-N-oxide 6a

A solution of 2.5 g (0.011 mol) of **9a** in 70 mL EtOH was hydrogenated at room temperature in the presence of 0.25 g 10% Pd/C. After hydrogen absorption ceased, the catalyst was filtered off and the filtrate was evaporated in vacuo. The solid residue was taken up with Et<sub>2</sub>O (15 mL) and filtered to give 1.93 g (yield 88%) of 4-amino-3-phenoxy-pyridine-*N*-oxide (**13a**), mp: 166–168 °C (benzene/EtOH 8:12). <sup>1</sup>H-NMR ( $d_6$ -DMSO)  $\delta$ : 6.1 (D<sub>2</sub>O exchangeable s, 1H, NH), 6.7 (d, 1H, J = 8 Hz, H-5), 7.0–7.3 (m, 3H), 7.3–7.6 (m, 2H), 7.6 (s, 1H, H-2), 7.8 (d, 1H, J = 8 Hz, H-6).

Conversion of **13a** into **6a** was accomplished following the procedure reported for **5a**. Yield 70%; mp: 199–200 °C (EtOH). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 2.2 (s, 3H, CH<sub>3</sub>), 6.8–7.2 (d, 2H, J = 10 Hz), 7.2–7.5 (m, 3H), 7.6–8.0 (m, 2H, H-2, H-5), 8.2 (D<sub>2</sub>O exchangeable s, 1H, NH), 8.4 (d, 1H, J = 8 Hz, H-6). Anal C<sub>13</sub>H<sub>12</sub>N<sub>2</sub>O<sub>3</sub> (C, H, N).

Pharmacological methods

Antihistaminic activity

Experiments were performed using male Hartley guinea pigs  $(400 \pm 45 \text{ g})$  body weight) obtained from BMG-Allevamento (Cividate al Piano, BG, Italy). The animals were housed under conditions of constant temperature and humidity with a standard diet (8-GP-17, Charles River, Italy) and water ad libitum. Fasted guinea pigs were anaesthetized (ethylurethane, 1.3-1.5 g/kg ip) and respiration was fully arrested with pancuronium bromide (2 mg/kg iv). Airway insufflation pressure (intratracheal pressure, ITP) and systemic blood pressure (BP), via femoral artery, were recorded following the method of

Konzett-Rössler using Hewlett-Packard instruments (Hewlett-Packard, Waltham, MA, USA) [1]. Bronchoconstriction was induced with histamine (10  $\mu$ g/kg iv). Test compounds were administered intravenously in a single dose of 10 mg/kg iv). Nimesulide (1 mg/kg iv), indomethacin (1mg/kg iv) and pyrilamine (2 mg/kg iv) were employed as reference standards.

# Antiinflammatory activity

Edema, induced by the subplantar injection of carrageenin in the hind paw of rats, was measured by the mercury displacement method [9]. Groups of five rats were used and the increase in paw volume was estimated over 6 h by subtracting the volume of the controlateral paw which received an equal volume of saline. Representative compounds 1a-4a were administered orally, 30 min before carrageenin injection, at a dose of 10 mg/kg. Nimesulide was administered at a dose of 4 mg/kg.

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